2021/TDC(CBCS)/EVEN/SEM/ CHMHCC-602T/055

TDC (CBCS) Even Semester Exam., September—2021

CHEMISTRY

(6th Semester)

Course No.: CHMHCC-602T

(Organic Chemistry)

Full Marks: 50
Pass Marks: 20

Time: 3 hours

The figures in the margin indicate full marks for the questions

SECTION—A

Answer any ten questions:

2×10=20

- 1. How can you distinguish between O CH₃—CH₃ and CH₃—OH by IR-spectroscopy?
- 2. State the electronic transitions that take place when an α , β -unsaturated carbonyl compound is irradiated by UV-radiation.

3. Calculate the λ_{max} value of the following compound :

- 4. How can the intermolecular and intramolecular H-bonding be distinguished by IR-spectroscopy?
- 5. What is chemical shift in NMR-spectroscopy?
- 6. Comment on the given δ -value of the methyl protons in the following compounds:

7. Comment on the number of signals and their splitting, if any, in the NMR-spectrum of the following compound:

- 8. How many NMR-signals would you expect from each of the following?
 - (i) CH₃—O—CH₃

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(Continued)

- 9. Give the structure of α and β -isomer of glucose in Fischer representation.
- **10.** What happens when glucose is treated with phenylhydrazine? Give equations also.
- 11. What are epimers? Give an example.
- 12. How would you show that sucrose is made up of a unit each of D-glucose and D-fructose and it does not contain any 'free' carbonyl group?
- 13. What do you mean by 'hypsochromic shift?'
- 14. What are the characteristics of a good dye?
- **15.** Write one synthesis of phenolphthalein. Mention the chromophore and auxochrome present in the dye.
- 16. What are mordant dyes? Give one example.
- Provide one method of synthesis and one application of polyester.
- 18. What is Buna-S rubber?

- 19. What is anionic polymerization? Why does the substrate in an anionic polymerization contain an electron withdrawing substituent?
- 20. What are elastomers and plasticizers?

SECTION-B

Answer any five questions:

6×5=30

2

2

2

- 21. (a) Which of an isolated carbonyl group and a conjugated carbonyl group absorbs at higher frequency? Justify your answer.
 - (b) Which out of the following compounds
 (I) and (II) is expected to show a lower
 C=O stretching frequency? Defend your answer:

 OCH_3 and OCH_3 (II)

(c) While the λ_{max} for 1,3-butadiene is observed at 217 nm, that of 1,3,5-hexatriene appears at 258 nm. Explain the observation.

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- 22. (a) Which region in the IR-spectra is called fingerprint region?(b) How will you distinguish between
 - the following pair of compounds by UV-visible spectroscopy?

2

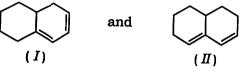
2

2

3

4

2



- (c) Explain, why the observed $n \to \pi^*$ absorption of acetone occurs at 279 nm in hexane, 270 nm in ethanol and 265 nm in water.
- 23. (a) Why is TMS used as a reference compound in NMR-spectroscopy? How does it help in chemical shift measurement? 2+1=3
 - (b) Draw a typical ¹H-NMR spectrum of ethanol explaining the chemical shift and splitting pattern of all the hydrogens involved.
- **24.** (a) Illustrate shielding and deshielding of protons taking ethyne and benzene as example.
 - (b) How can you distinguish between toluene and benzoic acid by ¹H-NMR spectroscopy taking chemical shift and splitting pattern in consideration?
- 22J**/129** (Turn Over)

22J/129

| 25. | (a) | reaction taking D-glucose as the starting material. 1+2=3 |
|-----|-------|---|
| | (b) | Convert D-glucose into gluconic acid. 1 |
| | (c) . | D-glucose and D-fructose form the same osazone. Explain. 2 |
| 26. | (a) | What is Kiliani-Fischer synthesis? Discuss the reaction taking D-arabinose as the starting material. 1+2=3 |
| | (b) | Represent the formula of cellulose as per Haworth projection. |
| 27. | (a) | Write one synthesis of methyl orange. Mention the chromophore and auxochrome present in the dye. 3+1=4 |
| | (b) | Explain bathochromic shift. 2 |
| 28. | (a) | Write one method of synthesis of malachite green. |
| | (b) | Discuss briefly the relationship between colour and constitution. |
| | | |

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(7)

| 29. | (a) | What is addition polymerization? | 1 |
|-----|-----|--|------|
| | (b) | Provide the mechanism of Ziegler-Natta catalysis. | 2½ |
| | (c) | Explain vulcanization of rubber. | 21/2 |
| 30. | (a) | Explain the mechanism of cationic polymerization reaction. | 3 |
| | (b) | What are isotactic, syndiotactic and atactic polymers? | 3 |
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